

Material Safety Data Sheet # 4001					Last Revision 09/21/12		Page 1 of 2		
		SECTIO	N 1: CHEM	ICAL PRODU	CT & COMPANY II	DENTIFICATION			
CHEMICAL NAME: Anhydrous Ammonia					TRADE NAMES / SYNONYMS: Ammonia				
DISTRIBUTOR:					EMERGENCY TELEPHONE NUMBERS:				
Airgas Specialty Products					Transportation (CH	1-800-424-9	1-800-424-9300		
2530 Sever Road, 300					Transportation, Canada (CANUTEC):): 1-613-996-6	1-613-996-6666	
Lawrenceville, GA 30043 USA					Environmental/Health/Safety (24-hr): 1-		1-800-528-4	963	
					Customer Service	1-800-295-2	225		
		SECTIO	ON 2: COM	POSITION / IN	FORMATION ON I	NGREDIENTS			
CHEMICAL	FORMULA	<u>% BY V</u>	VEIGHT	CAS	OSHA PEL	NIOSH REL / A	SH REL / ACGIH TLV IDLH		
		<u>C-grade</u>	P-grade		25 ppm (California only)				
Ammonia	NH_3	99.5	99.995	7664-41-7	50 ppm (TWA)	25 ppm (TWA)	35 ppm (STEL)	300ppm	
Water	H ₂ O	0.4	33 ppm	7732-18-5	None	None	None		
			-						

None

None

None

SECTION 3: HAZARDS IDENTIFICATION EMERGENCY OVERVIEW: 1. Colorless gas or compressed liquid with a pungent, suffocating odor **2.** Liquid ammonia reacts violently with water. Vapor cloud is produced. **3.** Avoid contact with liquid and vapor. **4.** Stay upwind and use water spray to absorb vapor. **5.** Not flammable under conditions likely to be encountered outdoors. **6.** Stop discharge if possible.

POTENTIAL HEALTH EFFECT

0.1

2 ppm

Oil

ROUTES OF ENTRY: Inhalation, Skin Contact, Eye Contact, Ingestion. **TARGET ORGANS:** Eyes, skin and respiratory system. **EYE CONTACT:** Exposure to liquid or high concentrations of vapor can cause painful, instant and possibly irreversible damage to tissue such as conjunctiva, cornea and lens. **SKIN CONTACT:** Prolonged contact with high concentrations can cause painful tissue damage, frostbite and serious chemical burns. **INHALATION:** Depending on exposure concentration and duration, effects can vary from none or only mild irritation, to obstruction of breathing from laryngeal and bronchial spasm, to edema and severe damage to mucous membranes of the respiratory tract with possible fatal results. Latent edema and residual reduction in pulmonary function may occur. **INGESTION:** Tissue damage, chemical burns, nausea and vomiting can occur. Ammonia is a gas under normal atmospheric conditions and ingestion is unlikely. **CARCINOGENICITY: NTP?** <u>No</u> **IARC?** <u>No</u> **OSHA?** <u>No</u>

SECTION 4: FIRST AID MEASURES

EYE CONTACT: Flush with large amounts of water for at least 15 minutes then immediately seek medical aid.

SKIN CONTACT: Immediately flush with large quantities of water for at least 15 minutes while removing clothing. If clothing has frozen to skin, thaw with water before removal. Seek immediate medical aid.

INHALATION: Remove from exposure. If breathing has stopped or is difficult, administer artificial respiration or oxygen as needed. Seek immediate medical aid.

INGESTION: Do not induce vomiting. Have victim drink large quantities of water if conscious. Immediately seek medical aid. Never give anything by mouth to an unconscious person.

SECTION 5: FIRE FIGHTING MEASURES

FLASH POINT(method used): Not Applicable **FLAMMABLE LIMITS:** 16-25% in air (for labeling purposes, not DOT flammable gas). **EXTINGUISHING MEDIA:** Stop flow of gas or liquid. Ammonia will burn in the range of 16-25% in air with a constant source of ignition. **SPECIAL FIRE FIGHTING PROCEDURES:** Move containers from fire zone if possible; if not, use water to cool fire-exposed containers. Use water spray to control vapors. Do not put water directly on liquid ammonia. Personnel must be equipped with appropriate protective clothing and respiratory protection.

NFPA HAZARD CLASSIFICATION:	Health: 3	Flammability: <u>1</u>	Reactivity: 0	(least-0 — 4-highest)
	SECTION 6:	ACCIDENTAL RELEASE	MEASURES	

In US, federal regulations require that a release of 100 lb. or more of ammonia must be reported immediately to the National Response Center at (800) 424-8802, the SERC and the LEPC. In California, ALL releases must be reported to CUPA, state and local agencies. Additional state and local regulations may apply. **SUGGESTED LOCAL ACTION:** Stop leak if feasible. Avoid breathing ammonia. Evacuate personnel not equipped with protective clothing and equipment. Use copious amounts of water spray or fog to absorb ammonia vapor. DO NOT put water on liquid ammonia. Contain run-off to prevent ammonia from entering a stream, lake, sewer, or ditch. Any release of this material, during the course of loading, transporting, unloading or temporary storage, must be reported to U.S. DOT as required by 49 CFR 171.15 and 171.16.

SECTION 7: HANDLING AND STORAGE

Refer to the ANSI K61.1 standard for storage and handling information. Protect containers from physical damage and temperatures exceeding 120°F. Use only approved storage systems. Zinc, copper, silver, cadmium, and their alloys must not be used in ammonia systems since they can be rapidly corroded by it. Avoid hydrostatic pressure, which can cause equipment rupture, by adhering to proper filling procedures and the use of hydrostatic pressure relief valves where appropriate.

SECTION 8: EXPOSURE CONTROLS/PERSONAL PROTECTION

RESPIRATORY PROTECTION: Respiratory protection approved by NIOSH / MSHA for ammonia must be used when exposure limits are exceeded. Whether chemical canister respirator or self-contained breathing apparatus is sufficient for effective respiratory protection depends on the type and magnitude of exposure.

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SKIN PROTECTION: Rubber gloves and rubber or other types of approved protective clothing should be used to prevent skin contact. A face shield should be used for increased protection from contact with liquid or vapor.

EYE PROTECTION: Chemical splash goggles, approved for use with ammonia, must be worn to prevent eye contact with liquid or vapor. A face shield should be used for increased protection from contact with liquid.

VENTILATION: Local positive pressure and/or exhaust ventilation should be used to reduce vapor concentrations in confined spaces. Ammonia vapor, being lighter than air, can be expected to dissipate to the upper atmosphere. Ammonia concentrations may also be reduced by the use of an appropriate absorbent or reactant material.

may also be reduced by	The use of all appropriate absor	bent of reactant material.			
	SECTION 9: PHYS	SICAL AND CHEMICAL PRO	DPERTIES		
BOILING POINT: -28.1	°F		TY: 0.62 @ 60°F (water=1)		
SOLUBILITY IN WATE	R: High	VAPOR DENSITY	VAPOR DENSITY: 0.60 @ 32°F (Air=1)		
MELTING POINT: -107			pH: Approx. 11.6 for 1 N Sol'n. in water		
PERCENT VOLATILE			Colorless, pungent gas		
VAPOR PRESSURE: 4	4802.9 mm Hg @ 60°F or 107.6	psia.			
	SECTION 10	: STABILITY AND REACTIN	VITY		
STABILITY: Material g	enerally considered stable. Hea	ting above ambient temperate	ure causes rapid increase of vapor pressure.		
			cids. Under certain conditions, ammonia		
			ontaneously. Reactions of ammonia with		
	o form explosive fulminate-like co				
			 The decomposition temperature may be 		
	ntact with certain metals such as				
HAZARDOUS POLYMI	ERIZATION: Will not occur		AVOID: Not applicable		
		TOXICOLOGICAL INFORM			
	cali and readily damages all body				
		enicy Effects: No informati	on is available and no adverse effects are		
anticipated. Synergistic	Materials: None known.				
		: ECOLOGICAL INFORMAT			
	2.0-2.5 ppm/1-4 days/ goldfish ar				
6	60-80 ppm/3 days/crayfish/LC ₁₀₀ ;	BIOCHEMI	CAL OXYGEN DEMAND: Not pertinent		
8			CONCENTRATION POTENTIAL: None		
		DISPOSAL CONSIDERATI			
			personnel experienced in ammonia spills		
			ncies for acceptable disposal procedures and		
disposal locations. For H	Hazardous Waste Regulations ca				
		4: TRANSPORT INFORMAT			
	DOMESTIC SHIPMENTS	INTERNATIONAL SHIPME			
Proper shipping name:		Ammonia, Anhydrous	Ammonia, Anhydrous		
Shipping Class:	DOT 2.2 (nonflammable gas)	2.3 (poison gas)	2.3 (8)		
Identification Number:	UN1005	UN1005	UN1005		
Packing Group:	None		None		
		REGULATORY INFORMA			
			tion 313 of Title III) and 40 CFR Part 370. Be		
	ly with state and local regulations		of opposite in any O4 hour pariod much ha		
			of ammonia in any 24-hour period must be ten follow-up is required to SERC & LEPC.		
	MUNICATION RULE, 20 CFR 19				
	CONTROL ACT: This material is				
			LE III): Section 302 Extremely Hazardous		
			Hazards; Section 313 Toxic Chemical: Yes.		
	(1%) CALIFORNIA PROPOSI		Carcinogen: No		
			ject to the Process Safety Management		
	R 1910.119 if maintained on-site	•	, , ,		
			product is subject to the Risk Management		
	CFR Part 68 if maintained on-si				
DRINKING WATER: M	laximum use dosage in potable v	water is 5mg/l.	-		
	SECTION	16: OTHER INFORMATION	N		
REASON FOR REVISIO			n Section 1. 2. Revised LEL and UEL from 16-		
			pecialty Products. 4. Canadian transportation		
			vised: 16-25%. 7. Company address changed.		
			Canadian TDG ACT under Section 14. MSDS		
PREPARED BY: Airgas					

This information is taken from sources or based upon data believed to be reliable, however, Airgas Specialty Products makes no warranty as to the absolute correctness or sufficiency of any of the foregoing or that additional or other measures may not be required under particular conditions.